

State Science Talent Search Examination, 2014

Time: 2 Hours Max. Marks: 150

Instructions to candidates:

- 1. Answer are to be marked on the separate Answer Sheet which will be collected after the time is over.
- Please write your Roll No. very clearly (only one digit in one block) as given on your Admission Card. 2. Please see that no block is left unfilled and even zeros appearing in the Roll No., if any, are correctly transferred to the appropriate blocks on the booklet and on the Answer Sheet.

Example:

R | J | 0 | 1 | 1 | 8 | 2

For all subsequent purposes, your Roll No. shall remain the same as given on the Admission Card.

- 3. This test is in SIX parts. Each part consists of 25 questions. In all, 150 questions are to be attempted. Each question carries one mark.
- Since all the questions are compulsory, do not try to read through the whole question paper before beginning 4. to answer it.
- 5. Begin with the first question and keep on attempting one question after another till you finish.
- 6. If you do not know the answer to any question, do not spend much time on it and pass on to the next one. If time permits, you can come back to the questions which you have left in the first instance and try them again.
- 7. Space has been provided for rough work at the bottom of each page.
- 8. Answer to each question is to be indicated by blackening, with H.B. pencil, the bubble (circle) of the correct alternative in the separately given Answer Sheet from amongst the ones given for the corresponding question in the test booklet. Please also read the instructions carefully, given on the back side of the Answer Sheet.

PHYSICS

(Question Nos. 1 to 25)

The distance travelled by a car whose speed is 35 kmh⁻¹ in 12 minutes will be 1.

$$(2) 7.0 \text{ km}$$

Ans.

Given: Speed = 35 kmh^{-1} Sol.

Time taken =
$$12 \text{ min} = \frac{12}{60} \text{ hr}$$

$$\because Speed = \frac{Distance}{Time}$$

$$\Rightarrow$$
 Distance = speed \times time

$$=35 \text{ kmh}^{-1} \times \frac{12}{60} \text{ h}$$

Distance = 7.0 km

- 2. The speed of a uniformly accelerated car changes from 18 kmh⁻¹ to 36 kmh⁻¹ in 5s. What is its acceleration in ms⁻²?
 - (1) 1.0
- (2) 1.5
- (3) 2.0
- (4) 3.0

Ans. [1]

Sol. Given: -

Initial speed = $18 \text{ kmh}^{-1} = 5 \text{ ms}^{-1}$

Final speed = $36 \text{ kmh}^{-1} = 10 \text{ ms}^{-1}$

Time taken = 5 sec

 $\therefore Acceleration (a) = \frac{change in speed}{time taken}$

$$a = \frac{v - u}{t} = \frac{10 - 5}{5}$$

 $a = 1.0 \text{ ms}^{-2}$

- 3. The motion of an object is said to be uniform circular motion if the object
 - (1) moves on a circular path with constant speed
 - (2) moves on a circular path with constant velocity
 - (3) moves on a circular path with variable speed
 - (4) none of these
- Ans. [1]
- **Sol.** The motion of an object is said to be uniform circular motion, if the object move on a circular path with constant speed.
- 4. A bullet of mass 10 g is horizontally fired with a velocity 150 ms⁻¹ from a pistol of mass 5 kg. The recoil velocity of the pistol will be
 - $(1) 0.5 \text{ ms}^{-1}$
- $(2) 0.4 \text{ ms}^{-1}$
- $(3) 0.3 \text{ ms}^{-1}$
- $(4) 0.2 \text{ ms}^{-1}$

- Ans. [3]
- **Sol.** Given:

Mass of bullet (m) = 10 gm

$$= 0.01 \text{ kg}$$

Velocity of gun (v) = 150 m/sec

Mass of pistol of mass (M) = 5 kg

$$\therefore$$
 Recoil velocity of the pistol is $\frac{mv}{M} = \frac{0.01 \times 150}{5} = \frac{1.5}{5} = 0.3$ m/sec

- 5. If A and B are two objects with masses 6 kg and 34 kg respectively then
 - (1) A has more inertia than B

- (2) B has more inertia than A
- (3) A and B have the same inertia
- (4) none of the two has inertia

- Ans. [2]
- **Sol.** B has more inertia than A. As inertia depends directly upon mass.



6.	The mass	of an	object is	s 5 ks	g. Its	weight or	n the	surface	of the	earth	will	be

- (1) 490 N
- (2) 0.49 N
- (3) 4.9 N
- (4) 49 N

Ans. **[4]**

Sol. Given:

Mass of an object = 5 kg

Therefore, weight = mg

 $=5\times9.8$

Weight = 49 N

7. SI unit of pressure is

- (1) pascal
- (2) newton
- (3) joule
- (4) watt

Ans. [1]

Sol. SI unit of pressure is pascal.

- Relative density of silver is 10.8. The density of water is 1 gm cm⁻³. Density of silver in SI unit will be 8.
 - (1) $10.8 \times 10^{-3} \text{ kg m}^{-3}$

(2) 10.8 kg m^{-3}

 $(3) 10^3 \text{ kg m}^{-3}$

(4) $10.8 \times 10^3 \text{ kg m}^{-3}$

Ans. [4]

Sol. Given:

Relative density of silver is 10.8

Density of water is 1 gm cm $^{-3}$ = 1000 kg m $^{-3}$

∴ Relative density of silver = Density of substance (silver)

Density of water

So. Density of silver = 10.8×1000

Density of silver = $10.8 \times 10^3 \text{ kg m}^{-3}$

- 9. Which physical quantity has kWh as its unit?
 - (1) Force
- (2) Momentum
- (3) Energy
- (4) Power

[3] Ans.

Sol. Energy unit is kWh.

- 10. A force of 12 N displaces a body by 60 cm in its direction. The work done on the body will be
 - (1) 720 J
- (2) 7.2 J
- (3) 5 J
- (4) 0.2 J

[2] Ans.

Sol. Given:

Displacement = 60 cm

= 0.6 m

Force applied (F) = 12 N

Work done = Force \times displacement

$$= 12 \times 0.6$$

$$W = 7.2 J$$



CAREER POINT

- 11. An object of mass 2 kg is moving with a constant velocity 2 ms⁻¹. How much work is needed to be done against the object in order to bring it to rest?
 - (1) 1 J
- (2) 4 J
- (3) 8 J
- (4) 10 J

Ans. [2]

Sol. According to work-energy theorem

W = change in kinetic energy

$$W = \frac{1}{2} mv^2 - \frac{1}{2} mu^2$$

$$v = 0 \text{ m/s}, \ u = 2 \text{ m/s}, \ m = 2 \text{ kg}$$

$$W = -\frac{1}{2} \times 2 \times (2)^2$$

$$W = -4 J$$

Therefore, work done against the object is 4 J.

- 12. The minimum time interval needed between the original sound and the reflected sound for hearing a distinct echo is
 - (1) 0.2 s
- (2) 2.0 s
- (3) 0.1 s
- (4) 1.0 s

Ans. [3]

Sol. We know that the persistence of sound is $\frac{1}{10}$ th second.

This means minimum time interval needed between the original sound & the reflected sound for hearing a distinct echo is 0.1 sec.

- 13. The audible range of sound for human beings is
 - (1) 2 Hz to 20 kHz

(2) 2 Hz to 2 kHz

(3) 20 Hz to 200 kHz

(4) 20 Hz to 20 kHz

Ans. [4]

- **Sol.** The audible range of sound for human beings is 20 Hz to 20 kHz.
- 14. A person claps his hands near a high tower and hears echo after 0.2 s. The speed of the sound is 300 ms⁻¹. The distance of the tower from the person is
 - (1) 30 m
- (2) 15 m
- (3) 6 m
- (4) 3 m

Ans. [1]

Sol. Given:

Time after echo is heard = 0.2 s

Speed of the sound (v) = 300 ms^{-1}

∵ we know that.

$$v = \frac{2d}{t} \implies 2d = v \times t$$

$$\Rightarrow$$
 d = $\frac{v \times t}{2} = \frac{300}{2} \times 0.2$

$$d = 30 \text{ m}$$

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CAREER POINT

- 15. The image formed by a concave mirror is of the same size, real and inverted when the object is placed
 - (1) at infinity
- (2) between P and F
- (3) at F
- (4) at C

- Ans. [4]
- The image formed by a concave mirror is the same size, real & inverted when the object is placed at C. Sol.
- 16. The focal length of a convex lens is 50 cm. Its power in dioptre is
 - (1) -2

- (4) + 1

- [3] Ans.
- Power = $\frac{1}{f(m)}$ Sol.

$$P = \frac{100}{f}$$
 Dioptre

For convex lens, focal length is positive

$$P = \frac{+100}{50} = +2$$

$$P = +2$$
 Dioptre

- The refractive index of glass is 1.5. If the speed of light in air is 3×10^8 ms⁻¹ then its speed in glass will be 17.
 - $(1) 2 \times 10^8 \text{ ms}^{-1}$
- (2) $3 \times 10^8 \text{ ms}^{-1}$ (3) $4.5 \times 10^8 \text{ ms}^{-1}$
- $(4) 6 \times 10^8 \text{ ms}^{-1}$

- [1] Ans.
- Sol. Given:

Refractive index of glass = 1.5

Speed of light in air = $3 \times 10^8 \text{ ms}^{-1}$

As we know.

$$_1\mu_2 = \frac{\mu_2}{\mu_1} \, = \, \frac{v_1}{v_2}$$

$$\Rightarrow v_2 = \frac{v_1}{\mu_2} \times \mu_1 = \frac{3 \times 10^8}{1.5} \times 1$$

speed of light in glass

$$v_2 = 2 \times 10^8 \text{ ms}^{-1}$$

- 18. The process of splitting white light into its seven constituent colours is called
 - (1) Refraction
- (2) Dispersion
- (3) Scattering
- (4) Reflection

- Ans.
- Sol. The process of splitting white light into its seven constituent colours is called dispersion.

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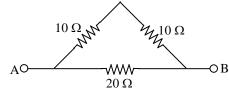
- 19. The danger indicators (signal) are red in colour because
 - (1) scattering of red light is maximum
- (2) scattering of red light is minimum
- (3) the wavelength of red light is minimum
- (4) none of these

Ans.

Sol. The danger indicators are red in colour because, scattering of red light is minimum.

Scattering
$$\propto \frac{1}{(\text{Wavelength})^4}$$

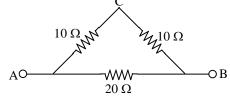
20. Equivalent resistance between points A and B in the given circuit will be



- $(1) 40 \Omega$
- (2) 30Ω
- $(3) 20 \Omega$
- (4) 10Ω

[4] Ans.

Sol.



AC and BC branches are connected in series.

i.e
$$R_{AC} + R_{BC}$$

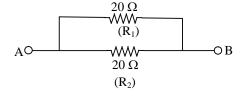
$$= 10 + 10 = 20 \Omega$$

$$\frac{1}{R_{eq}} = \frac{1}{R_1} + \frac{1}{R_2}$$

$$R_1=R_2=R=20\;\Omega$$

$$\therefore R_{eq} = \frac{R}{n}$$

$$R_{eq} = \frac{20}{2} = 10 \, \Omega$$



- 21. Which of the following terms represents electrical power in a circuit?
 - (1) I² Rt
- (2) VIt
- (3) $I^2 R$

[3] Ans.

Sol. : Power = VI
$$\Rightarrow$$
 P = I²R [: V = IR]

- 22. The electrical energy dissipated per second in a resistance of 4 Ω is 100 J. The current flowing through the resistance will be
 - (1) 25 A
- (2) 15 A
- (3) 10 A
- (4) 5 A

Ans. [4]

Sol. Given:

Time, t = 1 sec

Resistance, $R = 4 \Omega$

Energy, E = 100 J

As we know that

$$E = I^2Rt$$

$$\Rightarrow I^2 = \frac{E}{R.t} = \frac{100}{4 \times 1}$$

$$I^2 = 25 A$$

$$\Rightarrow$$
 I = 5A

- 23. The frequency of an alternating current is 50 Hz. In how much time does it reverse its direction?
 - (1) $\frac{1}{10}$ second
- (2) $\frac{1}{100}$ second
- (3) 10 second
- (4) 100 second

Ans. [2]

Sol. Given:

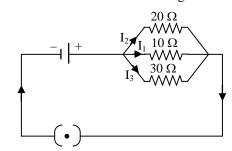
Frequency of an alternating current is 50 Hz.

⇒ time taken to reverse its direction

$$\Rightarrow t = \frac{1}{2f}$$

$$t = \frac{1}{100} \sec$$

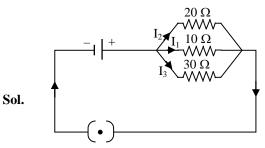
24. Which of the following statements is true on the basis of the given circuit diagram?



- (1) I_1 is maximum
- (2) I₂ is maximum
- (3) I₃ is maximum
- (4) $I_1 + I_2 + I_3 = 0$

Ans. [1]

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I₁ is maximum, because resistance is minimum and they all are connected is parallel. i.e. voltage remain constant.

$$\left(I \propto \frac{1}{R}\right)$$

- 25. The magnetic effect of current was discovered by
 - (1) Faraday
- (2) Millikan
- (3) Oersted
- (4) Thomson

Ans. [3]

Sol. The magnetic effect of current was discovered by Oersted.

CHEMISTRY

(Question Nos. 1 to 25)

- 1. The nuclei, which are not identical but have the same number of nucleons, are called -(1) isotopes (4) isoelectronic (2) isotones (3) isobars Ans. [3] Sol. **Isobars** Nucleons include both neutron & proton.
- 2. Aqueous solution of which of the following compounds turns red litmus to blue?
 - (1) CO₂
- (2) SO₂
- (3) MgO
- (4) SO₃

Ans. [3]

Sol. MgO on reaction with water forms basic solution, which turns red litmus blue.

- 3. The colloidal solution in which both dispersed phase and dispersion medium are liquids, is called -
 - (1) Gel
- (2) Sol
- (3) Aerosol
- (4) Emulsion

[4] Ans.

- 4. Aqueous solution of copper sulphate can be stored in metal vessel -
 - (1) Al
- (2) Zn
- (3) Fe
- (4) Ag

Ans. [4]

Sol. Because Ag is less reactive than Cu, whereas Al, Zn & Fe are more reactive, So displacement reaction will take place.

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- **5.** Which group of modern periodic table contains gas, liquid and solid non-metallic elements?
 - (1) 12
- (2) 13
- (3) 17
- (4) 18

Ans. [3]

Sol. Group 17

 $F, Cl, \rightarrow Gas$

 $Br \rightarrow Liquid$

I & At \rightarrow Solid

- **6.** Colloidal particles exhibit Tyndall effect due to -
 - (1) polarization of light (2) scattering of light
- (3) reflection of light
- (4) refraction of light

Ans. [2]

- **7.** Which is correct statement?
 - (1) sodium is non-reactive metal
 - (2) non-metals react with acids to give hydrogen
 - (3) copper and silver are malleable and ductile
 - (4) non-metals do not show allotropy

Ans. [3]

Sol. Cu & Ag are malleable as well as ductile.

- **8.** Cinnabar is ore -
 - (1) sulphide

- (2) carbonate
- (3) oxide
- (4) sulphate

Ans. [1]

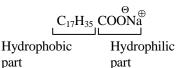
Sol. HgS Cinnabar

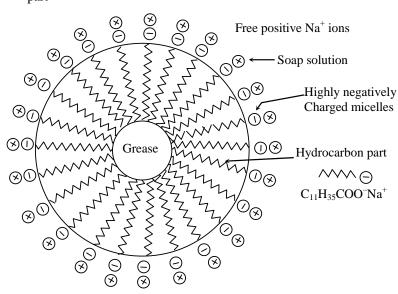
- **9.** Ethanol is changed to ethanoic acid by -
 - (1) oxidation reaction
 - (2) reduction reaction
 - (3) addition reaction
 - (4) substitution reaction

Ans. [1]

Sol. C_2H_5 -OH + 2[O] $\xrightarrow{\text{Alkaline KMnO}_4}$ $CH_3\text{COOH} + H_2\text{O}$

- 10. The action of soaps and detergents to remove the oily dirt by emulsification is due to -
 - (1) presence of hydrophilic group
 - (2) presence of hydrophobic group
 - (3) presence of both hydrophilic and hydrophobic groups
 - (4) presence of hardness in water
- Ans. [3]
- **Sol.** These part orient themselves to form micelle, which causes emulsification.





Cleansing action of soap. Soap micelle entraps the oily dirt particle

- 11. Which functional group cannot be situated at terminal position?
 - (1) carboxylic acid
- (2) aldehyde
- (3) alcohol
- (4) ketone

Ans. [4]

Sol. Ketone (–CO–)is a intermediate group R–CO–R

Whereas carboxylic acids (R-COOH), Aldehyde (R-CHO) & Alcohol (R-OH) are terminal function groups.

- **12.** Natural indicator is -
 - (1) methyl orange
- (2) phenolphthalein
- (3) litmus
- (4) methyl red

Ans. [3]

Sol. Litmus obtained from Lichen.

- 13. pH of soda water is-
 - (1)7
- (2) between 7 9
- (3) between 4-7
- (4) 14

[3] Ans.

- Because soda water is made by dissolving CO₂ in water & it is acidic by nature (H₂CO₃). Sol.
- 14. Number of atoms in 40 grams of He is -
 - $(1) 6.022 \times 10^{22}$
- $(2) 6.022 \times 10^{23}$
- $(3) 6.022 \times 10^{24}$
- $(4) 6.022 \times 10^{25}$

[3] Ans.

4g He = 1 mole = 6.022×10^{23} atoms Sol.

So 40 g He = 10 mole =
$$6.022 \times 10^{23} \times 10 = 6.022 \times 10^{24}$$
 atom

- **15.** Which metal cannot be extracted from its ore by reduction using carbon?
 - (1) Mg
- (2) Zn
- (3) Cu
- (4) Fe

Ans. [1]

- Sol. Zn, Cu & Fe are metals of medium reactivity, So they can be reduced by using C, where as Mg is a highly reactive metal so it can only be reduced by electrolysis method.
- 16. The substance that gives Bleaching powder on reaction with chlorine is -
 - (1) CaO
- (2) CaCO₃
- (3) Ca(OH)₂
- (4) CaCl₂

Ans. [3]

- Sol. $Ca(OH)_2 + Cl_2 \rightarrow CaOCl_2 + H_2O$
- 17. Number of covalent bonds in cyclohexane is -
 - (1) 14
- (2) 16
- (3) 18
- (4) 20

[3] Ans.

Sol.

- 18. The substance made up of only one type of element is -
 - (1) Common salt
- (2) Diamond
- (3) Water
- (4) Air

Ans. [2]

Sol. Common salt → compound

Water \rightarrow compound

 $Air \rightarrow Mixture$

Diamond \rightarrow Made of C only.

A metalloid with lowest atomic number belongs to group of periodic table -								
(1) 11	(2) 12	(3) 13	(4) 14					
[3]								
Boron								
Elements present in sar	ne group of period	lic table are similar with	respect to -					
(1) valence electrons	(2) atomic radi	us (3) number of	shells (4) mass number					
[1]								
All the members of a p	articular group of	periodic table are having	same valence electron as well as	valency				
Organic compound is -								
(1) CO ₂	(2) CaCO ₃	(3) Ca(HCO ₃)	$(4) (NH_2)_2CO$					
[4]								
CaCO ₃ & Ca(HCO ₃) ₂ ca	arries metal Ca, S	o they can't be organic.						
Isotope of carbon that is isobar of stable isotope of nitrogen and isotone of stable isotope of oxygen is -								
$(1)_{6}^{12}C$	$(2)_{6}^{13}C$	(3) $_{6}^{14}$ C	$(4)_{6}^{12}C^{+}$					
[3]								
C-14 contain $14 - 6 = 8$	8 neutrons, which	is isobar to N-14 & isoto	one to O-16.					
Which metal does not liberate hydrogen gas by reaction with H ₂ SO ₄ ?								
(1) Cu	(2) Zn	(3) Mg	(4) Fe					
[1]								
Out of Zn, Mg & Fe, or	nly Cu is less reac	tive than H ₂ . So Cu can't	displace H ₂ gas from H ₂ SO ₄ .					
ores are roa	asted to convert in	to oxides -						
(1) Sulphide	(2) Carbonate	(3) Oxide	(4) Halide					
[1]								
The allotropic form of	carbon that shows	conductivity is -						
(1) Coal	(2) Diamond	(3) Graphite	(4) Fullerene					
[3]								
In graphite one of the v	valence of C is not	satisfied, So it can cond	act electricity.					
	(1) 11 [3] Boron Elements present in san (1) valence electrons [1] All the members of a p Organic compound is - (1) CO ₂ [4] CaCO ₃ & Ca(HCO ₃) ₂ can Isotope of carbon that in (1) ½ C [3] C-14 contain 14 – 6 = 8 Which metal does not b (1) Cu [1] Out of Zn, Mg & Fe, or	[1] (2) 12 [3] Boron Elements present in same group of period (1) valence electrons (2) atomic radia [1] All the members of a particular group of Organic compound is - (1) CO ₂ (2) CaCO ₃ [4] CaCO ₃ & Ca(HCO ₃) ₂ carries metal Ca, So Isotope of carbon that is isobar of stable is (1) ¹² / ₆ C (2) ¹³ / ₆ C [3] C-14 contain 14 – 6 = 8 neutrons, which Which metal does not liberate hydrogen is (1) Cu (2) Zn [1] Out of Zn, Mg & Fe, only Cu is less reaction or same roasted to convert in (1) Sulphide (2) Carbonate [1] The allotropic form of carbon that shows (1) Coal (2) Diamond [3]	(1) 11 (2) 12 (3) 13 [3] Boron Elements present in same group of periodic table are similar with (1) valence electrons (2) atomic radius (3) number of [1] All the members of a particular group of periodic table are having Organic compound is - (1) CO ₂ (2) CaCO ₃ (3) Ca(HCO ₃); [4] CaCO ₃ & Ca(HCO ₃) ₂ carries metal Ca, So they can't be organic. Isotope of carbon that is isobar of stable isotope of nitrogen and is (1) ${}^{12}_{6}$ C (2) ${}^{13}_{6}$ C (3) ${}^{14}_{6}$ C [3] C-14 contain 14 – 6 = 8 neutrons, which is isobar to N-14 & isotometric which metal does not liberate hydrogen gas by reaction with H ₂ Si (1) Cu (2) Zn (3) Mg [1] Out of Zn, Mg & Fe, only Cu is less reactive than H ₂ . So Cu can't	(1) 11 (2) 12 (3) 13 (4) 14 [3] Boron Elements present in same group of periodic table are similar with respect to - (1) valence electrons (2) atomic radius (3) number of shells (4) mass number [1] All the members of a particular group of periodic table are having same valence electron as well as Organic compound is - (1) CO ₂ (2) CaCO ₃ (3) Ca(HCO ₃) ₂ (4) (NH ₂) ₂ CO [4] CaCO ₃ & Ca(HCO ₃) ₂ carries metal Ca, So they can't be organic. Isotope of carbon that is isobar of stable isotope of nitrogen and isotone of stable isotope of oxyger (1) $_{6}^{12}$ C (2) $_{6}^{13}$ C (3) $_{6}^{14}$ C (4) $_{6}^{12}$ C* [3] C-14 contain 14 – 6 = 8 neutrons, which is isobar to N-14 & isotone to O-16. Which metal does not liberate hydrogen gas by reaction with H ₂ SO ₄ ? (1) Cu (2) Zn (3) Mg (4) Fe [1] Out of Zn, Mg & Fe, only Cu is less reactive than H ₂ . So Cu can't displace H ₂ gas from H ₂ SO ₄ .				

BIOLOGY

(Question Nos. 1 to 25)

1.	The name of the scientist presenting cell theory is							
	(1) Robert Brown	(2) Leeuwenhoek	(3) Virchow	(4) Schleiden and Schwann				
Ans.	[4]							
Sol.	Schleiden and Schwani	1						
2.	The growth in the girth	of stem or root is due to	which tissue ?					
	(1) Lateral meristem	(2) Apical meristem	(3) Intercalary meristem	(4) Parenchyma				
Ans.	[1]							
Sol.	Lateral meristem							
3.	Which tissue is respons	sible for movement in our	body?					
	(1) Epithelial tissue	(2) Connective tissue	(3) Muscular tissue	(4) Nervous tissue				
Ans.	[3]							
Sol.	Muscular tissue							
4.	The example of gymno	sperm plant is -						
	(1) Marchantia	(2) Pinus	(3) Marsilea	(4) Chara				
Ans.	[2]							
Sol.	Pinus							
5.	Which animal has an o	pen circulatory system and	d kidney like organ for excı	retion ?				
	(1) Palaemon	(2) Planaria	(3) Ascaris	(4) Pila				
Ans.	[4]							
Sol.	Pila							
6.	Animals of which class	have hairs on the skin, sv	weat glands and oil glands?	,				
	(1) Mammalia	(2) Aves	(3) Amphibia	(4) Reptilia				
Ans.	[1]							
Sol.	Mammalia							
7.	Diseases caused by bac	teria are						
	(1) Influenza and dengi	ue fever	(2) AIDS and cholera					
	(3) Cholera and tubercu	ılosis	(4) Malaria and kala-aza	r				
Ans.	[3]							
Sol.	Cholera and tuberculos	is						
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8.	The gas responsible for depletion of the ozone layer is						
	(1) Chlorofluorocarbon	(2) Methane	(3) Carbon dioxide	(4) Sulphur dioxide			
Ans.	[1]						
Sol.	Chlorofluorocarbon						
9.	Kharif crop is						
	(1) Gram	(2) Wheat	(3) Pea	(4) Maize			
Ans.	[4]						
Sol.	Maize						
10.	Which animal is used fo	r preparing vermicomposi	t ?				
	(1) Hydra	(2) Earthworm	(3) Ascaris	(4) Frog			
Ans.	[2]						
Sol.	Earthworm						
11.	Which animal is cultivat	ed for the production of p	pearl?				
	(1) Oysters	(2) Mussels	(3) Prawn	(4) Mullets			
Ans.	[1]						
Sol.	Oysters						
12.	The variety of honeybee	used in the commercial p	production of honey is				
	(1) Apis cerana indica	(2) Apis dorsata	(3) Apis florae	(4) Apis mellifera			
Ans.	[4]						
Sol.	Apis mellifera						
13.	Pepsin acts as digestive	enzyme in which organ?					
	(1) Intestine	(2) Stomach	(3) Liver	(4) Mouth			
Ans.	[2]						
Sol.	Stomach						
14.	Breaking down of pyruv	rate takes place in which c	eell organism ?				
	(1) Ribosome	(2) Golgi bodies	(3) Lysosome	(4) Mitochondria			
Ans.	[4]						
Sol.	Mitochondria						

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The function of platelets is -

15.

SSTSE-2014 EXAMINATION



CAREER POINT

	(1) Transportation of O ₂		(2) Storage of food material		
	(3) Controlling blood	d pressure	(4) Clotting of blood		
Ans.	[4]				
Sol.	Clotting of blood				
16.	The plant tissue resp	onsible for movement of w	rater and minerals obtained	d from the soil is	
	(1) Parenchyma	(2) Phloem	(3) Xylem	(4) Collenchyma	
Ans.	[3]				
Sol.	Xylem				
17.	The organ that remo	ves nitrogenous waste prod	uct from blood is		
	(1) Kidney	(2) Liver	(3) Lungs	(4) Heart	
Ans.	[1]				
Sol.	Kidney				
18.	The thinking part of	the brain is			
	(1) Mid-brain	(2) Hindbrain	(3) Forebrain	(4) Spinal cord	
Ans.	[3]				
Sol.	Forebrain				
19.	'The growth of polle	n tube towards ovules' show	ws which phenomenon?		
	(1) Geotropism	(2) Chemotropism	(3) Phototropism	(4) Hydrotropism	
Ans.	[2]				
Sol.	Chemotropism				
20.	Presence of swollen	neck is the symptom of wh	ich disease ?		
	(1) Goitre	(2) Diabetes	(3) Pellagra	(4) Albinism	
Ans.	[1]				
Sol.	Goitre				
21.	Multiple fission type	e of reproduction is found in	1		
	(1) Earthworm	(2) Yeast	(3) Hydra	(4) Plasmodium	
Ans.	[4]				
Sol.	Plasmodium				

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- 22. The example of unisexual flower is
 - (1) Hibiscus
- (2) Papaya
- (3) Mustard
- (4) Pea

- Ans. [2]
- Sol. Papaya
- 23. The method of reproduction by which 'Leaf of Bryophyllum develops into new plant' is called
 - (1) Budding
- (2) Binary fission
- (3) Multiple fission
- (4) Vegetative reproduction

- Ans. [4]
- Sol. Vegetative reproduction
- 24. Genotypic ratio in monohybrid cross is
 - (1) 2 : 1 : 1
- (2) 3:1
- (3) 1:2:1
- (4) 1:3

- Ans. [3]
- Sol. 1:2:1
- 25. In evolution, the wings of birds and bats are evidences of
 - (1) Homologous organ (2) Analogous organ
- (3) Fossils
- (4) Embryo

- Ans. [2]
- Sol. Analogous organ

MATHEMATICS

(Question Nos. 1 to 25)

- 1. Which one is the largest number among the following?
 - $(1) \ 0.37\overline{5}$
- $(2) \ 0.\overline{375}$
- $(3) \ 0.3\overline{75}$
- (4) 0.375

- Ans. [3]
- $(i) = 0.3755\overline{5}$ Sol.
 - (ii) 0.375375375
 - (iii) 0.3757575
 - (iv) 0.37500
- If $x = \sqrt[3]{2\frac{93}{125}}$, then value of x is -2.
 - $(1) 1\frac{1}{5}$
- (2) $2\frac{1}{5}$
- (3) $1\frac{2}{5}$

- Ans.
- Sol.

$$\Rightarrow x = \sqrt[3]{\frac{343}{125}} = x = \sqrt[3]{\left(\frac{7}{5}\right)^3} \Rightarrow x = \frac{7}{5} = 1\frac{2}{5}$$

- If the polynomials $ax^3 + 3x^2 13$ and $2x^3 5x + a$ are divided by (x 2) and leave the same remainder then 3. the value of a is -
 - (1) 1
- (2)2
- (3) 1/2
- (4) 1/7

[1] Ans.

 $ax^3 + 3x^2 - 13$ Sol.

 $2x^{3} - 5x + a$

remainder theorem ⇒

put x = 2

Put $x = 2 \Rightarrow 8a + 12 - 13$

16 - 10 + a

- 8a + 12 13 = 16 10 + a
- $7a = 7 \implies a = 1$
- 4. The value of k for which the following pair of linear equations has no solution, is -

$$(k-1) x + 4y - 1 = 0$$

$$4x + 9(k + 1)y + 2 = 0$$

- (1) 4/3
- (2) 5/3
- (3) 16/9
- (4) 25/9

[2] Ans.

Sol. For no solution

$$\frac{a_1}{a_2} = \frac{b_1}{b_2} \neq \frac{C_1}{C_2}$$

$$a_1 = (k-1), b_1 = 4, C_1 = -1$$

$$a_2 = 4$$
, $b_2 = 9(k + 1)$, $C_2 = 2$

$$\frac{a_1}{a_2} = \frac{b_1}{b_2} \Longrightarrow \frac{k-1}{4} = \frac{4}{9(k+1)}$$

$$\Rightarrow$$
 (9k + 9)(k - 1) = 16

$$\Rightarrow 9k^2 - 9k + 9k - 9 = 16$$

$$\Rightarrow$$
 9k² = 25

$$k^2 = \frac{25}{9} \Rightarrow k = \frac{5}{3}$$

5. A quadratic equation whose one root is 7 and the sum of roots is 5, is -

$$(1) x^2 + 5x + 14 = 0$$

(2)
$$x^2 - 5x - 14 = 0$$

(2)
$$x^2 - 5x - 14 = 0$$
 (3) $x^2 + 5x - 14 = 0$ (4) $x^2 - 5x + 14 = 0$

(4)
$$x^2 - 5x + 14 = 0$$

Ans.

Sol. given $\alpha = 7$ and $\alpha + \beta = 5$

so
$$\beta = -2$$

$$\alpha\beta = 7x - 2 = -14$$

$$x^2$$
 – (sum of root) x + product of roots = 0

$$\Rightarrow$$
 x² - (5)x + (-14) = 0

$$x^2 - 5x - 14 = 0$$

- If $\frac{2}{3}$, k and $\frac{5}{8}$ k are three consecutive terms of an arithmetic progression, then the value of k is -
 - $(1) \frac{16}{33}$
- (2) $\frac{7}{11}$
- $(3) \frac{33}{16}$
- $(4) \frac{16}{22}$

Ans. [1]

Sol.
$$\frac{2}{3}$$
, k, $\frac{5}{8}$ k

By using A.P. Property

$$k - \frac{2}{3} = \frac{5}{8}k - k$$

$$\frac{3k-2}{3} = \frac{5k-8k}{8} \implies \frac{3k-2}{3} = \frac{-3k}{8}$$

$$\Rightarrow$$
 24 k - 16 = -9k

$$\Rightarrow$$
 33k = 16 \Rightarrow k = $\frac{16}{33}$

- The sum of first n terms of an arithmetic progression is $3n + n^2$, then common difference of this arithmetic 7. progression is -
 - (1) n + 1
- (2) n
- (3) 2
- (4) 2n + 1

[3] Ans.

Sol.
$$S_n = 3n + n^2$$

$$S_{(n-1)} = 3(n-1) + (n-1)^2 = n^2 + n - 2$$

$$S_{(n-2)} = 3(n-2) + (n-2)^2 = n^2 - n - 2$$

$$n \text{ term of } AP = S_n - S_{n-1} = 2n + 2$$

$$(n-1)$$
 term of $AP = S_{n-1} - S_{n-2} = 2n$

So common difference \Rightarrow n term – (n – 1) term

$$\Rightarrow$$
 2n + 2 - 2n = 2

- If $\cos 2x = \sin (x 39^\circ)$ and 3x is acute angle then value of x is 8.
 - $(1) 40^{\circ}$
- $(2) 29^{\circ}$
- $(3) 43^{\circ}$
- $(4) 90^{\circ}$

Ans. [3]

Sol. $\cos 2x = \sin(x - 39^{\circ})$ and 3x is acute angle

$$\cos 2x = \cos(90 - x + 39^{\circ})$$

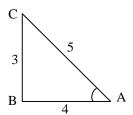
$$2x = 90 - x + 39$$

$$3x = 129^{\circ} \Rightarrow x = 43^{\circ}$$

- If $5 \sin A = 3$ and $0 < A < 90^{\circ}$ then value of (sec A + tan A) $(1 \sin A)$ is 9.
 - (1) 3/4
- (2) 4/5
- (3) 3/5
- (4) 2/5

- Ans. [2]
- Sol. $5\sin A = 3$

$$\sin A = \frac{3}{5} = \frac{P}{H}$$



$$\Rightarrow$$
 sec A = $\frac{1}{\cos A} = \frac{H}{B} = \frac{5}{4}$

$$\tan A = \frac{3}{4}$$

Put in (sec $A + \tan A$) $(1 - \sin A)$

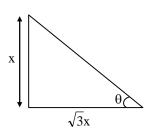
$$\Rightarrow \left(\frac{5}{4} + \frac{3}{4}\right)\left(1 - \frac{3}{5}\right)$$

$$\Rightarrow \frac{8}{4} \times \frac{2}{5} = \frac{8}{10} = \frac{4}{5}$$

- Length of shadow of a pole is $\sqrt{3}$ times the height of a pole. Angle of elevation of sun is = 10.
 - $(1) 30^{\circ}$
- $(2) 60^{\circ}$
- (3) 45°
- $(4)75^{\circ}$

- Ans.
- [1]

Sol.



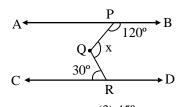
Shadow = $\sqrt{3}$ Height = $\sqrt{3}$ x

Angle of elevation of $\sin \theta = \tan \theta = \frac{x}{\sqrt{3}x}$

$$\Rightarrow \tan \theta = \frac{1}{\sqrt{3}}$$

$$\Rightarrow \theta = 30^{\circ}$$

In the following figure, AB||CD; \angle BPQ = 120° and \angle CRQ = 30°. The value of x is -11.

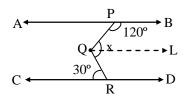


- $(1) 150^{\circ}$
- $(2) 60^{\circ}$

 $(4) 90^{\circ}$

Ans. [4]

Sol. Cons: - Draw LQ || BA



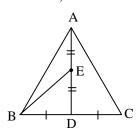
- \Rightarrow \angle BPQ + \angle PQL = 180° (Co-interior angles sum)
- $\Rightarrow \angle POL = 180^{\circ} 120^{\circ} = 60^{\circ}$

Similarly, $\angle RQL = \angle QRC = 30^{\circ}$ (alternate angles)

- $\therefore \angle POR = x^{\circ} = 60^{\circ} + 30^{\circ} = 90^{\circ}$
- 12. In triangle ABC, E is the mid-point of median AD. The correct statement among the following is -
 - (1) Area ($\triangle BED$) = $\frac{1}{4}$ Area($\triangle ABC$)
- (2) Area ($\triangle BED$) = $\frac{1}{4}$ Area ($\triangle ABD$)
- (3) $\frac{3}{4}$ Area (\triangle ABC) = Area (\triangle BED)
- (4) Area (\triangle ABC) = 2 Area (\triangle BED)

Ans. [1]

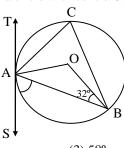
In ΔABD, Sol.



E is mid point of AD

- ∴ BE is median of ∆ABD
- ∴ Area of $\triangle BED = \frac{1}{2}$ Area of $\triangle ABD$
- $\therefore \text{ Area of } \Delta BED = \frac{1}{4} \text{ Area of } \Delta ABC$

13. In the given figure, TAS is a tangent to a circle with centre O. If \angle OBA = 32° then value of \angle BAS is -



- $(1)48^{\circ}$
- $(2) 32^{\circ}$
- $(3) 58^{\circ}$
- $(4) 90^{\circ}$

Ans.

Sol. $\angle BAS = \angle ACB$ (angle made by tangent & chord in one segment is equal to angle made by same chord on another segment)

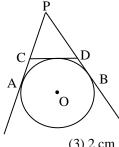
$$\angle$$
OBA = \angle OAB = 32°

$$(:: OA = OB)$$

$$\angle$$
 AOB = $180^{\circ} - 2 \times 32^{\circ}$
= $180^{\circ} - 64^{\circ}$

$$\angle ACB = \frac{1}{2} \angle AOB = \frac{1}{2} \times 116^{\circ} = 58^{\circ}$$

14. In the given figure, two tangents PA and PB are drawn to a circle with centre O from an external point P. CD is the third tangent touching the circle at Q. If PB = 10 cm and CQ = 2 cm, then length of PC is -

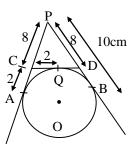


- (1) 6 cm
- (2) 5 cm
- (3) 2 cm
- (4) 8 cm

Ans.

[4]

Sol.



Length of PB = PA = 10 cm (Tangents drawn from an external Point are equal)

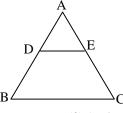
$$AC = CQ = 2 \text{ cm}$$

So,
$$PC = PA - AC$$

$$= 10 - 2$$

$$= 8 \text{ cm}$$

15. In the given figure,, DE \parallel BC and AD : DB = 2 : 3, then Area (\triangle ADE) : Area (\triangle ABC) is -



- (1) 4:25
- (2) 4:9
- (3) 9:4
- (4) 25:4

- Ans. [1]
- Sol. $\frac{AD}{DB} = \frac{2}{3}$

 \triangle ADE ~ \triangle ABC (By AA Similarity)

 $\angle DAE = \angle BAC$ (common)

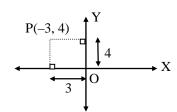
 $\angle ADE = \angle ABC$ (Corresponding angles)

Area ($\triangle ADE$): Area ($\triangle ABC$) = $\frac{(AD)^2}{(AB)^2} = \frac{(2)^2}{(5)^2} = \frac{4}{25}$

- \Rightarrow 4:25
- 16. A point P(-3, 4) lies in XY-plane. Perpendicular distance of the point P from Y-axis is -
 - (1) 3
- (2)4
- (3) 4
- (4) 3

Ans. [4]

Sol.



Perpendicular distance of point P from y - axis = 3

- (Distance is +ve)
- 17. The coordinate of the point on Y-axis which is equidistant from the points (6, 5) and (-4, 3) is -
 - (1)(0,0)
- (2)(0,9)
- (3)(9,0)
- (4)(1,4)

Ans. [2]

Sol. Let point be (0, b)

$$\sqrt{(6)^2 + (b-5)^2} = \sqrt{(4)^2 + (b-3)^2}$$

$$\Rightarrow 36 + b^2 + 25 - 10b = 16 + b^2 + 9 - 6b$$

$$\Rightarrow 36 - 10b + 6b = 0$$

$$\Rightarrow 36 - 4b = 0$$

$$\Rightarrow$$
 b = 9

point (0, 9)

- 18. Rain water from a roof of dimension $22 \text{ m} \times 20 \text{ m}$ drains into a cylindrical vessel having diameter of base 2 m and height 3.5 cm. If the vessel is just full, then the total rainfall in cm is -
 - (1) 3.5
- (2) 2.5
- (3) 2
- (4) 1.5

Ans. [Bonus]

Sol. Let total rainfall in cm be h cm

Then, diameter of base = 2 m = 200 cm

Radius = 100 cm

$$\Rightarrow l \times b \times h = \pi r^2 h_{cy} \cdot vessel$$

$$\Rightarrow 2200 \times 20 \times 100 \times h = \frac{22}{7} \times (100 \times 100) \times 3.5$$

$$\Rightarrow h = \frac{22}{7} \times \frac{3.5}{22 \times 20}$$

$$=\frac{0.5}{20}=\frac{1}{40}=0.025$$
 cm

- 19. The perimeter of a sector of a circle of radius 5.2 cm is 16.4 cm. Then area of the sector is -
 - $(1) 80 \text{ cm}^2$
- (2) 21.6 cm
- $(3) 15.6 \text{ cm}^2$
- $(4) 156 \text{ cm}^2$

Ans. [3]

Sol. Perimeter of sector = $\frac{\theta}{360^{\circ}} \times 2\pi r + 2r$

$$\Rightarrow 16.4 = \frac{\theta}{360^{\circ}} \times 2\pi r + 2r$$

$$\Rightarrow \frac{16.4 - 2r}{2\pi r} = \frac{\theta}{360^{\circ}}$$

Area of sector =
$$\frac{\theta}{360^{\circ}} \times \pi r^2$$

$$\Rightarrow \frac{16.4 - 2r}{2\pi r} \times \pi r^2$$

$$\Rightarrow \frac{8.2-r}{r} \times r^2$$

$$\Rightarrow$$
 (8.2 – r) × r

$$\Rightarrow$$
 (8.2 – 5.2) \times 5.2

$$\Rightarrow$$
 3 × 5.2

 \Rightarrow 15.6 cm²

- 20. Three metallic spheres of radii 3 cm, 4 cm and 5 cm respectively are melted to form a cone of radius 6 cm. Then the height of this cone is -
 - (1) 24 cm
- (2) 42 cm
- (3) 60 cm
- (4) 18 cm

Ans. [1]

Sol. Volume of three spheres = Volume of cone

$$\frac{4}{3}\pi r_1^3 + \frac{4}{3}\pi r_2^3 + \frac{4}{3}\pi r_3^3 = \frac{1}{3}\pi r_{cone}^2 h$$

$$\Rightarrow \frac{4}{3}\pi(r_1^3 + r_2^3 + r_3^3) = \frac{1}{3}\pi r_{cone}^2 \times h$$

$$\Rightarrow \frac{4}{3}(3^3 + 4^3 + 5^3) = \frac{1}{3} \times 6^2 \times h$$

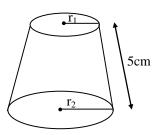
$$\Rightarrow$$
 4(27 + 64 + 125) = 36 h

$$\Rightarrow h = \frac{216}{9} = 24cm$$

- 21. The slant height of a frustum of a cone is 5 cm and the circumferences of its circular ends are 12π cm and 6π cm. Height of this frustum of cone is
 - (1) 3 cm
- (2) 4 cm
- (3) 5 cm
- (4) 6 cm

Ans. [2]

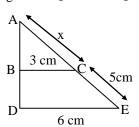
Sol.



Given $2\pi r_1 = 6\pi$

$$r_1 = 3$$
 cm

given
$$2\pi r_2 = 12\pi \Rightarrow r_2 = 6$$
 cm



$$\Rightarrow \frac{x}{3} = \frac{x+5}{6}$$

$$\Rightarrow$$
 x = 5 cm

Pythagoras in ΔABC

$$\Rightarrow$$
 AB = 4cm

Pythagoras in ΔADE

$$AD = 8 \text{ cm}$$

So height =
$$AD - AB$$

$$= (8-4) \text{ cm} = 4 \text{ cm}$$

22. The mean of the following data is 3.2. Then the value of x is -

X	1 – 3	3 – 5	5 – 7	7 – 9	9 – 11
Frequency	7	8	X	2	1

(1)6

(2) 3

(3) 2

(4) 4

[Bonus] Ans.

Sol.

Xi	f	Mid value (x)	f(x)
1 – 3	7	2	14
3 - 5	8	4	32
5 - 7	X	6	6x
7 - 9	2	8	16
9 – 11	1	10	10
	$\Sigma f = 18 + x$		$\Sigma f(x) = 72 + 6x$

Mean(
$$\bar{x}$$
) = $\frac{\Sigma fx}{\Sigma f}$ = 3.2 (given)

$$\frac{72+6x}{18+x} = \frac{32}{10}$$

$$\Rightarrow$$
 720 + 60x = 576 + 32x

$$\Rightarrow$$
 28x = 576 – 720

$$\Rightarrow x = \frac{-144}{28} = -5.1$$

23. If the median of the following data written in ascending order is 16, then value of x is -

$$6, 7, x - 2, x, 17, 20$$

- (1) 16
- (2) 18
- (3) 17
- (4) 15

[3] Ans.

Sol. Median =
$$\frac{x-2+x}{2}$$
 = 16 (given)

$$\frac{2x-2}{2} = 16 \implies x-1 = 16 \implies x = 17$$

- 24. A box contains 20 cards which are numbered from 1 to 20. A card is drawn at random from the box. Then the probability that it is a perfect square number is -
 - (1) 2/5
- (2) 1/20
- (3) 1/5
- (4) 4/5

Ans. [3]

Sol. Perfect square from 1 to 20

$$= 1, 4, 9, 16 = m$$

Probability =
$$\frac{m}{n} = \frac{4}{20} = \frac{1}{5}$$

- 25. Two dice are thrown at the same time. The probability that the sum of the numbers appearing on the top of the two dice is 9, is -
 - (1) 9/36
- (2) 1/6
- (3) 2/9
- (4) 1/9

Ans. [4]

Sol. Favorable cases = (3, 6), (4, 5), (5, 4), (6, 3) = m = 4

sample space
$$\Rightarrow$$
 n = (36)

Probability =
$$\frac{m}{3} = \frac{4}{36} = \frac{1}{9}$$

DAY TO DAY SCIENCE

(Question Nos. 1 to 25)

- 1. In which of the following speed of sound is maximum?
 - (1) Steel
- (2) Water
- (3) Hydrogen
- (4) Air

Ans. [1]

- Sol. Speed of sound is maximum in steel.
- 2. Which is known as the 'Country of winds'?
 - (1) Germany
- (2) India
- (3) Denmark
- (4) Nepal

Ans. [3]

- Sol. The 'country of winds' is Denmark.
- 3. The conventional unit for domestic electrical consumption is
 - (1) kWh
- (2) kWs
- (3) Ws
- (4) J

[1] Ans.

Sol. The conventional unit for domestic electrical consumption is kWh



CAREER POINT

4.	•	ncreasing the temperature		(4) 27 (6.1		
A	(1) Decreases	(2) Increases	(3) remains unchanged	(4) None of these		
Ans. Sol.	[2]	narassina tha tamparatura	of the medium increases.			
501.	The speed of sound on h	ncreasing the temperature	of the medium increases.			
5.	The acceleration of a fre	ely falling body is				
	$(1) + 9.8 \text{ ms}^{-2}$	$(2) - 9.8 \text{ ms}^{-2}$	(3) Not fixed	$(4) 0 \text{ ms}^{-2}$		
Ans.	[1]					
Sol.	The acceleration of a fre	ely falling body is $+g = +$	9.8 m/sec			
6.	Thermosetting polymer	is				
	(1) Terylene	(2) PVC	(3) Bakelite	(4) Nylon 6.6		
Ans.	[3]					
Sol.	Bakelite					
7.	Which medicine is not a					
	(1) Sodium hydrogen ca	rbonate	(2) Aluminium hydroxide			
	(3) Milk of magnesia		(4) Aspirin			
Ans.	[4]					
Sol.	All other are basic by na	ture so they can be used a	as antacid.			
	·	•				
8.	Natural rubber is					
	(1) Polyisoprin	(2) Neoprin	(3) Buna-S	(4) Buna-N		
Ans.	[1]					
Sol.	Polyisoprin is prepared	by isoprene (2-methyl-1,3	3-butadiene). Which is fou	and in latex, obtained from rubber		
	trees.					
9.	The feater responsible for	on o aid main is				
9.	The factor responsible for (1) Vaporisation of pollu		(2) Presence of frame in	atmagnhara		
	(3) Presence of oxides of		(4) Greenhouse effect.	(2) Presence of freons in atmosphere		
Ans.	[3]	i C, N, S in aunosphere	(4) Offeelinouse effect.			
Sol.		ing of fossil fuels, which	have CN&S forms	oxides of these elements, are the		
501.	major sources of acid rai	=	i have c, iv & 5, forms	oxides of these elements, are the		
	major boaroos or dord rd.	(302, 1102 & 302)				
10.	The particles bombarded	d to cause nuclear fission a	are			
	(1) Neutron	(2) Deuteron	(3) Alpha particle	(4) Beta particle		
Ans.	[1]					

Slow speed neutrons are bombarded on uranium 235 in nuclear fission

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Sol.

- 11. Which is not a natural fibre?
 - (1) Cotton
- (2) Silk
- (3) Wool
- (4) Dacron

[4] Ans.

Sol. Dacron

Cotton obtained from plant, silk & wool obtained from animals.

12. Consider the following statements:

If p is a prime number such that p + 2 is also a prime number, then

- I. p(p+2) + 1 is a perfect square number.
- II. 12 is a divisor of p + (p + 2), if p > 3

Which of the above statement (s) is / are correct?

- (1) Only I
- (2) Only II
- (3) Both I and II
- (4) Neither I nor II

Ans. [3]

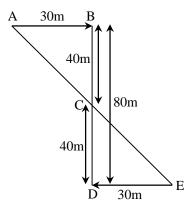
- 13. The ratio of A to B is x: 8 and the ratio of B to C is 12: y. If the ration of A to C is 2: 1, then the ratio of x to y is -
 - (1) 3 : 4
- (2) 4:3
- (3) 1 : 6
- (4) 1 : 12

Ans. [2]

- 14. Amit walked 30 m towards east, took a right turn and walked 80m, then he took a left turn and walked 30 m. The shortest distance between his starting position and present position is
 - (1) 30 m
- (2) 60 m
- (3) 90 m
- (4) 100 m

Ans. [4]

Sol.



The shortest distance will be from point A (starting point) to point E (final point)

As ΔABC is right angled triangle

$$AC = \sqrt{AB^2 + BC^2}$$

$$= \sqrt{(30)^2 + (40)^2}$$

$$= \sqrt{2500}$$

$$AC = 50 \text{ m}$$

Similarly, CE = 50 m

Total distance of A to E = 100 m

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CAREER POINT

15.	From your pocket money, you have to save Re 1 on first day, Rs. 2 on second day, Rs 3 on third day and on. Total money that will be saved in the month of October, 2014 is					
Ans.	(1) Rs. 496 [1]	(2) Rs. 500	(3) Rs. 992	(4) Rs. 31		
16.	A is twice as good a wo can A alone finish the wo	ork ?		rk in 14 days. In how many days		
Ans.	(1) 7 days [3]	(2) 18 days	(3) 21 days	(4) 28 days		
17.	In binary system number (1) 35	101101 is written in deci (2) 45	mal system as (3) 40	(4) 50		
Ans.	[2]	`,	` '			
18.	The scientist who prepare		(2) II D	(A) Weter and Crists		
Ans. Sol.	(1) Dr. Ian Wilmut [1] Dr. Ian Wilmut	(2) S.Cohen	(3) H.Boyer	(4) Watson and Crick		
19.	In which marine algae m (1) Dillisk	aximum quantity of iodin (2) Spirogyra	e is found? (3) Kelp	(4) Ulothrix		
Ans. Sol.	[3] Kelp	(=) = F== 85,==	(-)	(), (),		
20.	Example of totipotent ce (1) Nerve cells	lls in a living being is (2) Blastomeres	(3) Epithelial cell	(4) Muscle cells		
Ans. Sol.	[2] Blastomeres	(2) Bustomeres	(3) Epitheriai ceri	(4) Muscle cens		
21.	Which bacteria is used to (1) E.coli	o obtain human insulin ? (2) Vibrio cholerae	(3) Salmonella typhi	(4) Mycobacterium leprae		
Ans. Sol.	[1] E.coli					
22.	The technique used for re (1) Biotechnology	emoval of brain tumor is (2) Tissue culture	(3) Laser rays	(4) Brachy therapy		
Ans. Sol.	[3] Laser rays	(2) Hissue culture	(3) Easer rays	(4) Bluelly dictupy		
23.	Highest amount of which (1) Carbohydrate	n organic compound is fou (2) Protein	and in Soyabean? (3) Fat	(4) Vitamin		
Ans. Sol.	[2] Protein	(2) 1100011	(3) 1 41	(4) Vitaliilii		
24.	The structure that transfe (1) Cloning	ers DNA from one animal (2) Explant	to another is known as (3) Interferon	(4) Vector		
Ans. Sol.	[4] Vector	(-, -	(-,	(1) 10000		
25.	The technique applied fo (1) Chemotherapy	r testing position and acti (2) CT scan	vities of different structure (3) X-ray	es in human body is (4) Biopsy		
Ans. Sol.	[2] CT scan	(2) 01 30	(0) 11 143	(., 2.opo)		

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General Knowledge Regarding Science (Question Nos. 1 to 25)

- 1. Weight of an object when we move it from pole to the equator -
 - (1) increases
- (2) decreases
- (3) remains constant
- (4) none of these

Ans.

Sol. Weight of an object which we move it from pole to the equator is decreases.

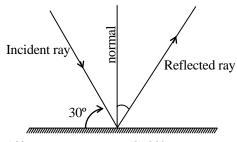
- 2. The orbit of each planet of the solar system is -
 - (1) circular
- (2) parabolic
- (3) elliptical

(4) rectangular

Ans. [3]

Sol. The orbit of each planet of the solar system is elliptical.

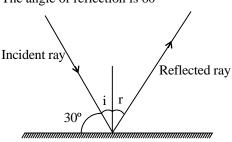
3. The angle of reflection in the figure is –



- $(1) 60^{\circ}$
- $(2) 90^{\circ}$
- $(3) 30^{\circ}$
- (4) None of these

Ans. [1]

Sol. The angle of reflection is 60°



$$\angle i + 30^{\circ} = 90^{\circ}$$

$$\angle i = 60^{\circ}$$

$$\angle i = \angle r$$
 (law of reflection).

$$\angle r = 60^{\circ}$$

- 4. Which lens is used to remove near-sightedness?
 - (1) Convex
- (2) Planoconvex
- (3) Planoconcave
- (4) Concave

[4] Ans.

Sol. Concave lens is used to remove near sightedness.

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CAREER POINT

- 5. Household electrical appliances are connected in parallel so that -
 - (1) uniform current flows

- (2) uniform voltage is received
- (3) energy consumption is high
- (4) none of these

Ans. [2]

- Sol. Household electrical appliances are connected in parallel so that uniform voltage is received
- 6. The process on which the energy produced in the sun is based, is -
 - (1) Nuclear fusion
- (2) Nuclear fission
- (3) Pair production
- (4) None of these

Ans. [1]

- Sol. The process on which the energy produced in the sun is based, is nuclear fusion.
- 7. When an object is dipped in a liquid the force of buoyancy on the object is -
 - (1) equal to the weight of the object
 - (2) equal to the weight of the displaced liquid
 - (3) equal to the twice the weight of the displaced liquid
 - (4) none of these

Ans.

- Sol. When an object is dipped in a liquid the force of buoyancy on the object is equal to the weight of the displaced liquid.
- 8. Hydrocarbon having highest octane number is -
 - (1) Isohexane
- (2) n-hexane
- (3) Iso-octane
- (4) n-octane

[3] Ans.

- 9. The chemical formula related to soap is -
 - (1) C₂H₅COONa
 - (2) CH₃(CH₂)₁₀CH₂OSO₃Na
 - (3) CH₃(CH₂)₁₅CH₂COONa
 - $(4) CH_3(CH_2)_{11}$

Ans. [3]

Sol. $CH_3(CH_2)_{15}CH_2COONa$

OR

C₁₇H₃₅COONa

(1) CO₂

The organic compound used as fire extinguisher is -

(2) C_6H_6

10.

SSTSE-2014 EXAMINATION

(3) COCl₂

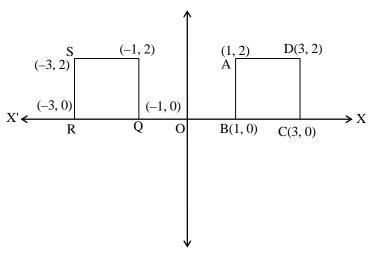


(4) CCl₄

CAREER POINT

Ans.	[4]							
11.	The role of added gypsum during manufacture of cement is to -							
	(1) facilitate gel for	mation	(2) decrease rate of	setting				
	(3) increase the wei	ght of product	(4) make the cemen	t impervious				
Ans.	[2]							
Sol.	If Gypsum is not ad	ded, the cement will set im	mediately leaving no tim	ne for concrete placing.				
12.	A fuel having highe	est thermal value is -						
	(1) Hydrogen	(2) Petrol	(3) LPG	(4) CNG				
Ans.	[1]							
Sol.	Out of Petrol, LPG	& CNG, H ₂ is having high	est calorific value around	d 150 kcal/kg.				
13.	The property not pro	esent in chlorofuluorocarbo	on compound is -					
	(1) non-toxic	(2) non-corrosive	(3) Volatility	(4) inflammatory				
Ans.	[4]							
14.	A child was born or	n Friday. Ist October in a ce	ortain year. His age on W	ednesday. 1st October, 2014 was -				
	(1) 2 years	(2) 5 years (3) 4	4 years (4)	6 years				
Ans.	[3]							
15.	If the bacteria of a	laboratory jar doubles eve	eryday and the whole jar	if filled with beteria in 30 days, then				
	number of days requ	uired to fill $\frac{1}{4}$ th of the jar i	S-					
	(1) 14 days	(2) 28 days	(3) 30 days	(4) 20 days				
Ans.	[2]	•	·	·				
16.	A motor boat goes	upstream on a river and o	coves the distance between	een two towns on the river bank in 6				
	hours. It covers this	distance downstream in 5	hours. If the sped of the	stream is 2 km/hour, then the speed of				
	motor boat in still w	vater will be -						
	(1) 20 km/hour	(2) 22 km/hour	(3) 29 km/hour	(4) 25 km/hour				
Ans.	[2]							

17. If in the given figure Y-axis works as a plane mirror then the image of point S(-3, 2) is –



- (1) A(1, 2)
- (2) B(1, 0)
- (3) D(3, 2)
- (4) C(3, 0)

Ans. [3]

- **18.** Producers in food chain are -
 - (1) Human
- (2) Plant
- (3) Frog
- (4) Insect

Ans. [2]

Sol. Plant

- **19.** Mutation borne disease is -
 - (1) Cancer
- (2) Haemophilia
- (3) AIDS
- (4) Colour blindness

Ans. [1]

Sol. Cancer

- **20.** Disease caused by the deficiency of protein is -
 - (1) Beri-beri
- (2) Scurvy
- (3) Kwashiorkor
- (4) Rickets

Ans. [3]

Sol. Kwashiorkor

- 21. World AIDS Day is celebrated on -
 - (1) 7th July
- (2) 1st December
- (3) 7th April
- (4) 5th May

Ans. [2]

Sol. 1st December

09	November, 2014	SSTSE-20°	14 EXAMINATION	P	CAREER POINT
22.	The method used for	preservation of liquid foo	odstuff is -		
	(1) canning	(2) dehydration	(3) pasteurization	(4) salt	solution

Ans. [3] Sol. Pasteurization 'Grey revolution' is related to which production area? 23. (1) Fertilizer (2) Milk (3) Potato (4) Fish Ans. [1] Sol. Fertilizer 24. The field related with Amrita Devi Bishnoi National Award is -(2) Energy conservation (1) Human conservation (3) Water conservation (4) Forest and wild life conservation Ans. [4] 25. In India the largest wind energy farm is established at -

In India the largest wind energy farm is established at (1) Mumbai
(2) Bengaluru
(3) Kanyakumari
(4) Ahmedabad
Ans. [3]

Sol.

Kanyakumari

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